## LEVEL – 2

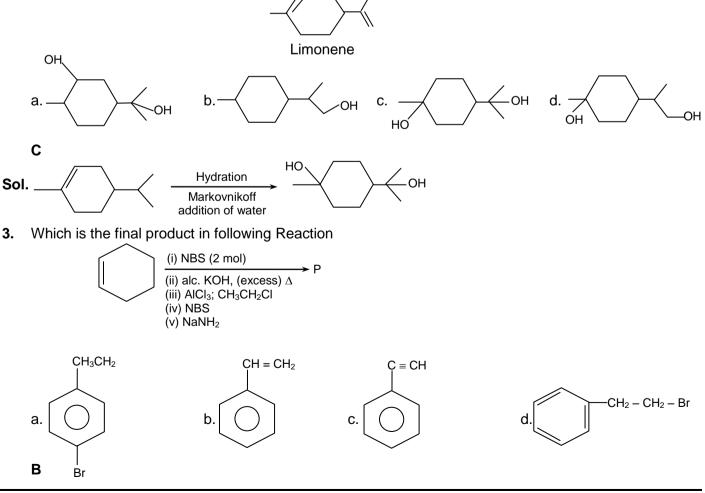
- 1. Total Questions: 40
- Time: 1 hour 15 minutes
- 2. Maximum Marks: 180
- 3. To be answered on other OMR.

SECTION – A (Single Correct Choice Type) Negative Marking [-1]

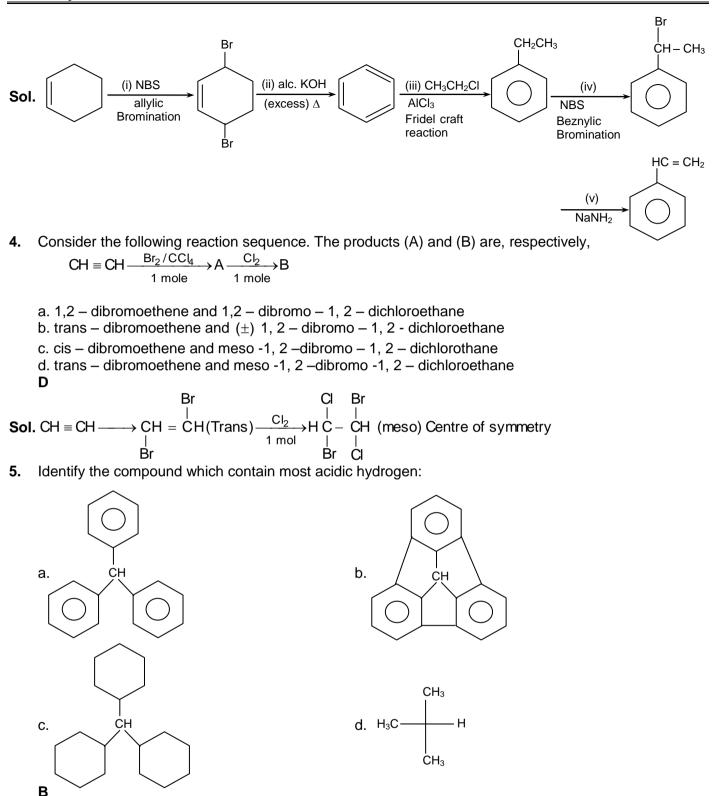
This Section contains **8 multiple choice questions.** Each question has four choices A), B), C) and D) out of which **ONLY ONE** is correct. (Mark only One choice) **8 × 4 = 32 Marks** 

- 1. 1.42 g of methyl iodide was completely converted into methyl magnesium iodide and the product was decomposed by excess of ethanol. How much volume of the gaseous hydrocarbon will be produced in the reaction at STP?
  - a. 22.4 litre b. 22400 mL C
- c. 0.224 litre
- d. 2.24 litre

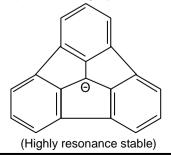
- Sol.  $CH_3I \longrightarrow CH_3MgI \longrightarrow CH_4$ 1 mol mole of  $CH_3I = \frac{1.42}{142} = 0.01$ mole of  $CH_4 = 0.01$ Volume = 0.01 × 22.4 lit = 0.224 lit
- 2. What product results from the complete Hydration of limonene, (in acidic medium) the structure of which is given below:



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**Sol.** Anion of the compound is highly stable as all three rings are in same plane.



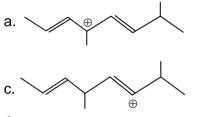


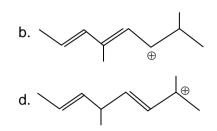
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- 6. In the Kjeldahl's method for estimation of nitrogen present in a soil sample, ammonia evolved from 0.75 gm of sample neutralized 10 ml of 1 M H<sub>2</sub>SO<sub>4</sub>. The percentage of nitrogen in the soil is
  - a. 37% b. 45% c. 73% d. 41%

Α

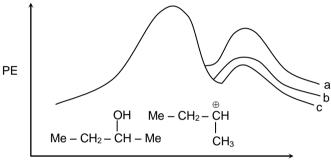
- **Sol.**% N =  $\frac{N_1V_1}{10} \times \frac{14}{W} = \frac{1 \times 2 \times 10}{10} \times \frac{15}{0.75} = 37\%$
- 7. Which carbocation is the most stable?





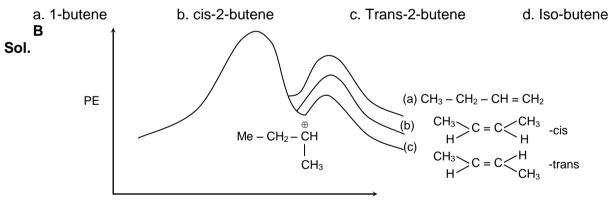
#### Α

- **Sol.** Stability of  $C^{\oplus} \propto$  extent of conjugation
- **8.** Energy profile diagram for dehydration of 2-butanol using conc.  $H_2SO_4$  is given below:



Reaction co-ordinate

Product (b) of above reaction is



Hence (a) is 1-butene Reaction co-ordinate b is cis-2-butene & c is trans 2-butene

# SECTION – B (Assertion and Reason) Negative Marking [-1]

This Section contains **10 multiple choice questions.** Each question has four choices A), B), C) and D) out of which **ONLY ONE** is correct. **10 × 4 = 40 Marks)** 

- (a) If both assertion and reason are true and reason is the correct explanation of assertion.
- (b) If both assertion and reason are true and reason is not correct explanation of assertion
- (c) If assertion is correct but reason is incorrect
- (d) If assertion is incorrect but reason is correct
- **1.** Assertion: 3° carbocation [(CH<sub>3</sub>)<sub>3</sub> $\overset{\oplus}{C}$ ] is more stable than 2° carbocation [(CH<sub>3</sub>)<sub>2</sub> $\overset{\oplus}{C}$ H] and 2°  $\oplus$

carbocation is more stable than the 1° carbocation ( $CH_3CH_2$ ).

**Reason:** Greater the number of hyperconjugative structures, more is the stability of carbocation.

a. (a) b. (b) c. (c) d. (d) A

Sol. Both assertion and reason are correct and reason is the correct explanation of assertion.

2. Assertion: Cyclopentadienyl anion is much more stable than allyl anion.

Reason: Cyclopentadienyl anion is aromatic in nature.

a. (a) b. (b) c. (c) d. (d)  
A 
$$\odot$$
  
Sol.  $=$   $\bigcirc$   $6\pi e^-$  (aromatic)

**3.** Assertion: Nitration of Benzene followed by Chlorination with Cl<sub>2</sub> & FeCl<sub>3</sub> give meta chloronitrobenzene. Reason: Nitro group on benzene will deactivate Benzene for electrophilic attack.

В

Sol. Nitro group is meta directing

4. Assertion (A): In Duma's method, organic compound is heated with CuO and then with Cu. Reason (R): It is done to avoid the oxidation of organic nitrogen into nitrogen oxides.

A Sol. It is method of estimation of nitrogen. Cu is used to convert nitrogen oxide to N<sub>2</sub>

Assertion(A): Cyclohexane –1, 2 –diol does not show geometrical isomerism.
 Reason (R): It has two chiral carbon.

Sol. Cyclohexane-1,2-diol has two geometrical isomer; cis = meso; Trans-d/ $\ell$ 

- 6. Assertion (A): Propyne is more acidic than propene. Reason (R): The conjugate base of propyne  $(CH_3 - C \equiv C^{\Theta})$  is more stable than that of propene  $\stackrel{\Theta}{}_{(CH_3 - CH = CH)}$ .
  - a. (a) b. (b) c. (c) d. (d) A

Sol. -ve charge on more E.N atom is more stable

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7. Assertion (A): The intermediate formed in the addition of HCI on 1-butene in the presence of benzoyl peroxide is:

 $CH_3 - CH_2 - CH_3 - CH_3$ 

**Reason (R) :** The H – Cl bond strength is higher than H – Br (which shows peroxide effect) and also H - Cl bond is heterolytically broken & have no peroxide effect

a. (a) b. (b) c. (c) d. (d)

8. Assertion (A): Addition of bromine to trans-2-butene yields meso-2, 3-dibromobutane. Reason (R) : Bromine addition to an alkene is a syn addition.

a. (a) <b>C</b>	b. (b)	c. (c)	d. (d)
C			

Sol. Addition of Halogen is anti addition

**9.** Assertion (A): cis 2-Butene on Hydroxylation by Baeyer's reagent forms optically inactive 2,3-Butanediol. Reason (R): Bayer's reagent is 1% alkaline potassium permagnate.

BÒ

Sol. Baeyer's Reagent is used for syn Hydroxylation; cis will form meso diol.

Assertion (A): Tropylium carbocation is more stable than benzyl carbonium ion.
 Reason (R): Tropylium carbocation maintains its aromatic character in all of its resonating structure but benzyl carbocation loses it aromatic character in some of the cannonical forms.

a. (a)	b. (b)	c. (c)	d. (d)
Α			. ,

## SECTION – C (Paragraph Type) Negative Marking [-1]

This Section contains **2 paragraphs.** Each of these questions has four choices A), B), C) and D) out of which **ONLY ONE** is correct. **8 × 4 = 32 Marks** 

#### Comprehension -1

Ozonolysis is one of the important method of determination of olefinic bonds in a compound. It is a cleavage reaction in which double bond is completely broken and alkene molecule is converted into two or more smaller molecules depending on the number of olefinic bonds with a carbonyl group. Ozonolysis is carried out in two stages. First is addition of ozone to give ozonide second step involve hydrolysis of ozonide in presence of reducing agent (usually zinc) to give product of reductive ozonolysis. In presence of zinc it yields aldehydes and ketones while presence of Ag<sub>2</sub>O, H<sub>2</sub>O or peracids form acids and/ or ketones. Further ozonides of alkynes yield carboxylic acid by cleavage through diketone which is oxidized. Acetylene give however mixture of glyoxal as well as formic acid.

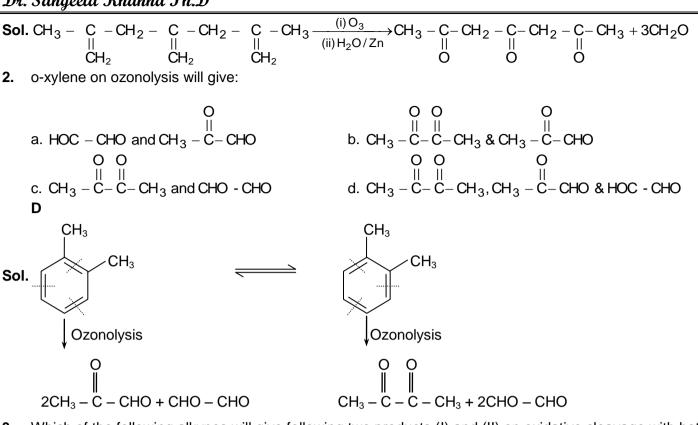
1. Consider the following reaction: (A)  $\xrightarrow{(i) O_3}_{(ii) H_2 O/Zn}$  CH<sub>3</sub>  $\xrightarrow{O}_{-}$  CH<sub>2</sub>  $\xrightarrow{O}_{-}$  H<sub>2</sub>  $\xrightarrow{O}_{+}$  + 3CH<sub>2</sub>O H<sub>3</sub>C-C-CH<sub>2</sub>  $\xrightarrow{I}_{-}$  H<sub>3</sub>C-C-CH<sub>2</sub>

a. 
$$CH_3 - C - CH_2 - C - CH_2 - C - CH_2 - C - CH_3$$
  
 $CH_2 - CH_2 - C - CH_2 - C - CH_3$   
c.  $CH_3 - C - CH_2 - C - CH_2 - C = CH_2$   
 $CH_3 - C - CH_2 - C - CH_2 - C = CH_2$ 

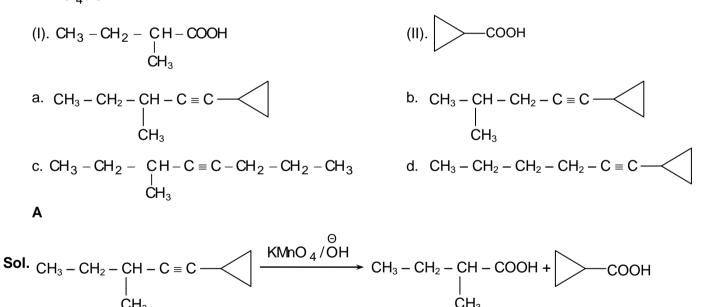
b. 
$$CH_3 - C - CH_2 - C - CH_2 - C - CH_3$$

d. all of these

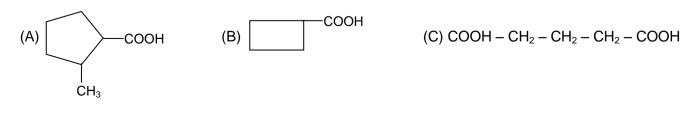
С

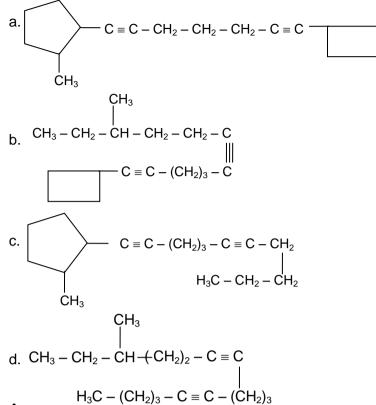


3. Which of the following alkynes will give following two products (I) and (II) on oxidative cleavage with hot  $_{O}^{O}$  KMnO<sub>4</sub>/OH?

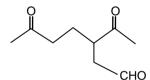


4. Which of the following alkynes will give following three products (A), (B) and (C) on oxidative cleavage  $\stackrel{\Theta}{}_{O}$  by hot KMnO<sub>4</sub>/OH?

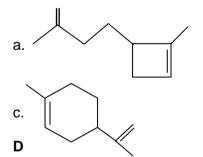


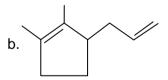


- Α
- **5.** Compound (A) on hydrogenation consumed 2 mol. equivalents of hydrogen and produces a saturated hydrocarbon. When A is treated with ozone, followed by reductive work up, two products are isolated : formaldehyde and the diketone structure of which is given below:



Compound (A) can be represented as :



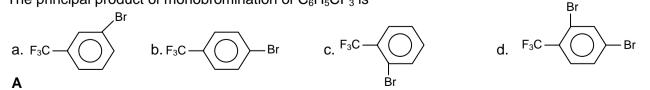


d. All of these

### Comprehension – 2

The 5 ring H's of monosubstituted benzenes  $C_6H_5G$  are not equally reactive. During the introduction of E into  $C_6H_5G$ , the 'G' determines the orientation of E. An electron donating G is o, p director, whereas an electron withdrawing G is meta director. Most of the meta directors are ring deactivating groups

6. The principal product of monobromination of  $C_6H_5CF_3$  is

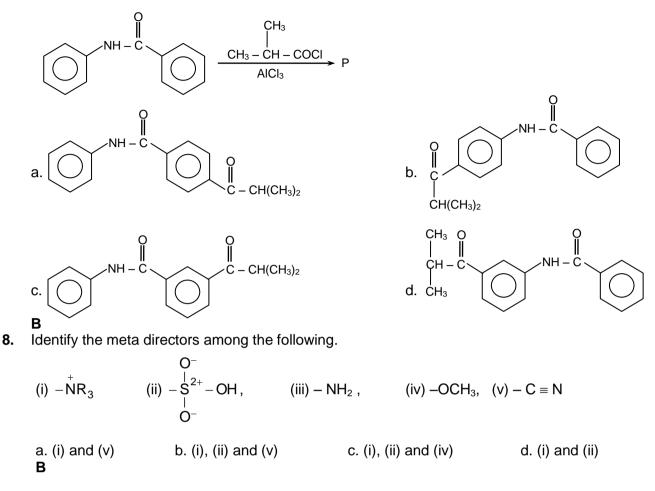


**Sol.** – CF<sub>3</sub> group is meta directing due to reverse Hyperconjugation

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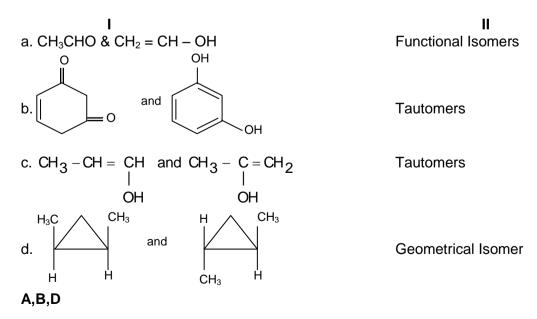
7. What will be the major product formed in following reaction



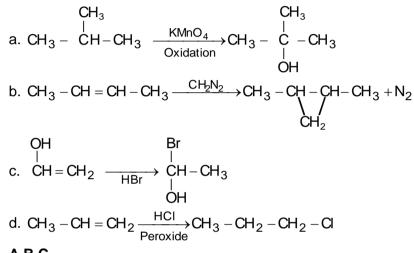
## SECTION - D (More than One Answer) No Negative Marking

This Section contains **6 questions.** Each question has four choices A), B), C) and D) out of which **ONE OR MORE** may be correct. (6 × 5 = 30 Marks)

1. Find correct Match



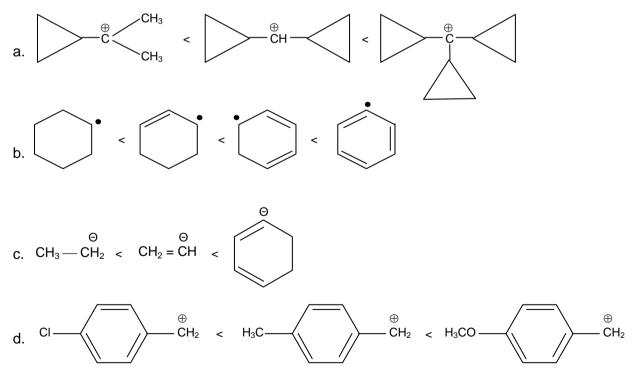
- 2. Which of the following will not give methane
- a. Decarboxylation of ethanoic acid c. Reduction of ethanol with Red P and HI **B**,C **Sol.**  $CH_3COOH \xrightarrow{Soda lime} CH_4 + CO_2$ ;  $CH_3CI \xrightarrow{Na}_{dry \ ether} CH_3CH_3$ ;  $C_2H_5OH \xrightarrow{RedP}_{HI} CH_3CH_3$  (d)  $CH_3MgBr \xrightarrow{HOCH_3} CH_4 + Mg \Big<_{OCH_3}^{Br}$
- 3. Which of the following reaction product is/are correct



A,B,C

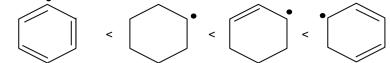
Sol. Antimarkovnikoff's Rule is not observed with HCI

4. Which of the following order of stability is correct?

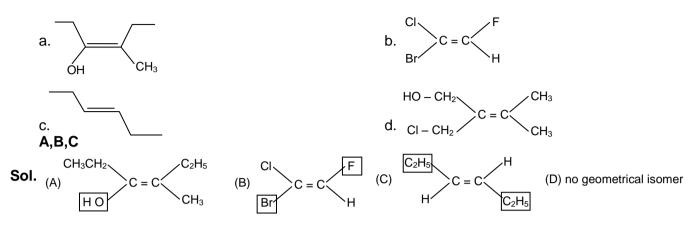


### A,C,D Sol. Correct order is

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5. Which of the following geometrical isomer have E-configuration.



6. Which of the following compounds have two chiral centre

a. Pentane-2,3-diol c. 2-Chloro-3-methylpentane **A,C** 

**Sol.** (1) 
$$CH_3 - CH - CH - CH_2 - CH_3$$
 (2)  
OH OH  
CH\_3  
(3)  $CH_3 - CH - CH_2 - CH_3$  (4)

b. 2,3-Dimethylpentane d. 1,4-Dichlorocyclohexane

$$\begin{array}{c} \mathsf{CH}_3 & \mathsf{CH}_3 \\ | & | \\ \mathsf{CH}_3 - \mathsf{CH} - \mathsf{CH}_3 - \mathsf{CH}_2 - \mathsf{CH}_3 \end{array}$$

(4) no chiral centre

# SECTION – E (Matrix Type) No Negative Marking

This Section contains **2 questions.** Each question has four choices (A, B, C and D) given in **Column I** and five statements (p, q, r, and s) in **Column II.** (2 × 8 = 16 Marks)

	Match the column (I) with column (II). One or more than One Column (I) Reaction		Column (II) Nature of products formed	
(a)	$CH_{3}$ $CH_{3} - CH_{2} - CH_{2} - CH_{2} \xrightarrow{H_{2}O/H^{+}} Pr oduct$ $CH_{3}$	(p)	Product formed are according of Markovni koff's Rule	
(b)	$CH_{3} - CH_{3} - CH_{2} - CH_{3} \xrightarrow[]{Alc.}{KOH} Pr oduct$	(q)	Rearrangement takes place	
(c)	$CH_{3} = CH_{3} - CH_{2} - CH_{2} \xrightarrow{CH_{3}}{I} CH_{3} - O^{\Theta}K^{+} \xrightarrow{CH_{3}}{I} O^{\Theta}K^{+} O^{\Theta}K^{+} \xrightarrow{CH_{3}}{I} O^{\Theta}K^{+} O^$	(r)	Major Product are formed according to Hoffman rule	
(d)	$\begin{array}{c c} & H_2SO_4 \\ \hline & & \Delta \end{array}  \text{Product} \\ \hline \\ & CH_3 \end{array}$	(s)	Major Products follow sytzeff's rule	

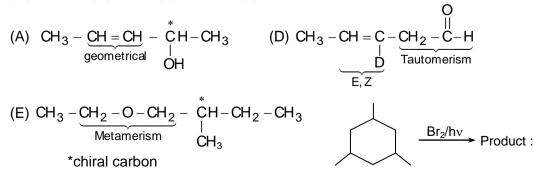
# 1. Match the column (I) with column (II). One or more than One Match

Sol. A  $\rightarrow$  p, q B  $\rightarrow$  s, C  $\rightarrow$  r, D  $\rightarrow$  q,s

2. In the following matrix some organic Compounds are given in Column-A which have to be matched with type of isomerism in Column-B. One or more than One Match

Column – I Compounds			Column – II Type of Isomerism can be shown		
(A)	$CH_3 - CH = CH - CH - CH_3$ $\downarrow$ OH	(P)	Geometrical isomerism		
(B)	CIFC = CBrI	(Q)	Tautomerism		
(C)	$CH_3 - CH = C - CH_2 - C - H$	(R)	Metamerism		
(D)	$\begin{array}{c} CH_3 - CH_2 - O - CH_2 - CH - CH_3 \\ \\ \\ C H_2 \\ \\ H_3 \end{array}$	(S)	Optical isomerism		

Sol.  $(A) \rightarrow P, S; (B) \rightarrow P; (C) \rightarrow P, Q; (D) \rightarrow R, S$ 

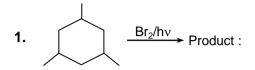


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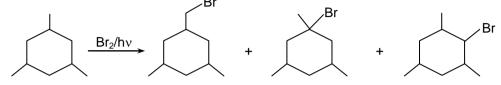
## SECTION – F (Integer Type) No Negative Marking

This Section contains 6 Questions. The answer to each question is a Single Digit Integer ranging from 0 to 10. The correct digit below the question number in the OMR is to be bubbled.

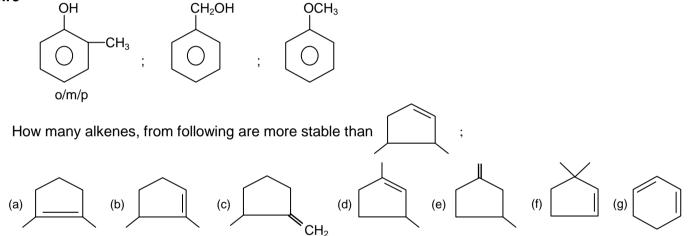
6 × 5 = 30 Marks



How many monobrominated products will be obtained by above reaction? (only structural) Sol.3



2. How many aromatic, isomers with Benzene are possible for  $C_7H_8O$  molecular formula Sol.5



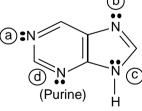
#### Sol.5

3.

a, b, d, e, g

g is a conjugated diene (more stable) & a, b, d & e have more Hyperconjugation.

The purine hetrocycle occurs mainly in the structure of DNA. Identify number of 'N' atoms having 4. localized lone pair of electron.



- Sol. 3; a, b, d (In C; lone pair is in conjugation)
- How many of the following do not have chain Isomer. 5.
  - C<sub>2</sub>H<sub>6</sub>, C<sub>3</sub>H<sub>6</sub>, C<sub>4</sub>H<sub>6</sub>, C<sub>4</sub>H<sub>8</sub>, C<sub>3</sub>H<sub>8</sub>, C<sub>2</sub>H<sub>2</sub>, C<sub>3</sub>H<sub>8</sub>O, C<sub>6</sub>H<sub>14</sub>,
- **Sol. 5**; C<sub>2</sub>H<sub>6</sub>; C<sub>3</sub>H<sub>6</sub>; C<sub>3</sub>H<sub>8</sub>, C<sub>3</sub>H<sub>8</sub>O; C<sub>2</sub>H<sub>2</sub>. Chain isomer can exist for compound with 4 carbon.
- The number of  $\pi$ -bonds in the product formed by passing acetylene through dilute sulphuric acid 6. containing mercuric sulphate is:

Sol. 1

$$CH \equiv CH \xrightarrow{HOH/Hg^{+2}} CH_2 = CH \xrightarrow{HOH_3 - CH_3} CH_3 - CH_3 -$$

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